

Putting Into Practice

Mass on balance - 1g precision:

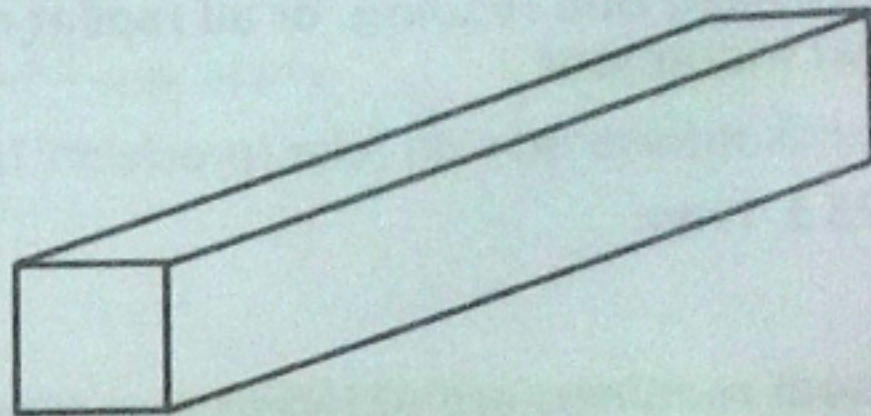
471g

Length with ruler - 1mm precision:

300mm, 300mm, 300mm

Width with vernier caliper - 0.1mm precision:

14.9mm, 14.6mm, 14.5mm, 14.4mm



1. Calculate the absolute and percentage uncertainty in the mass measurement.

$$\text{absolute} = \pm 1 \text{ g}$$

$$\begin{aligned} \% &= \frac{1}{471} \times 100 \\ &= 0.212\% \end{aligned}$$

2. Calculate the average length and its absolute and percentage uncertainty.

$$\text{average} = 300 \text{ mm}$$

$$\text{absolute} = \pm 1 \text{ mm}$$

$$\begin{aligned} \% &= \frac{1}{300} \times 100 \\ &= 0.33\% \end{aligned}$$

3. Calculate the average width and its absolute and percentage uncertainty.

$$\begin{aligned} \text{average} &= 14.9 + 14.6 + 14.5 + 14.4 \\ &= 14.6 \text{ mm} \end{aligned}$$

$$\text{absolute} = \frac{0.5}{2} = 0.25 \text{ mm}$$

$$\% = \frac{0.25}{14.6} \times 100 = 1.71\%$$

4. Calculate the cross sectional area of the square face and its percentage uncertainty.

$$14.6^2 = 213.16 \text{ mm}^2$$

$$1.71 \times 2 = 3.42\%$$

5. The formula for the density of the rod is: density = mass / (area x length)

What will be the percentage error in the calculated density? (No need to actually calculate the density.)

$$0.33 + 1.71$$

$$0.33 + 3.42 + 0.212 = 3.962$$

$$= 4\%$$